

Bulliant FT Success #1: Phase Sep'n

$$\Delta G_{mix} = RT (m_1 \ln \phi_1 + m_2 \ln \phi_2 + X m_1 \phi_2)$$

$$\frac{\partial}{\partial m_1} \Delta G_{mix} = \mu_1 - \mu_1^o$$

Need: $\frac{\partial \phi_1}{\partial m_1} = \frac{\partial}{\partial m_1} \left[\frac{m_1}{m_1 + \sigma m_2} \right] = \frac{\phi_2}{m_1 + \sigma m_2}$

$$\frac{\partial \phi_2}{\partial m_1} = \frac{\partial}{\partial m_1} \left[\frac{\sigma m_2}{m_1 + \sigma m_2} \right] = \frac{-\phi_2}{m_1 + \sigma m_2}$$

Ask them to prove it

★

The result is: $\mu_1 - \mu_1^o = RT \left[\ln(1 - \phi_2) + \phi_2 \left(1 - \frac{\sigma}{\tau}\right) + X \phi_2^2 \right]$

This equation

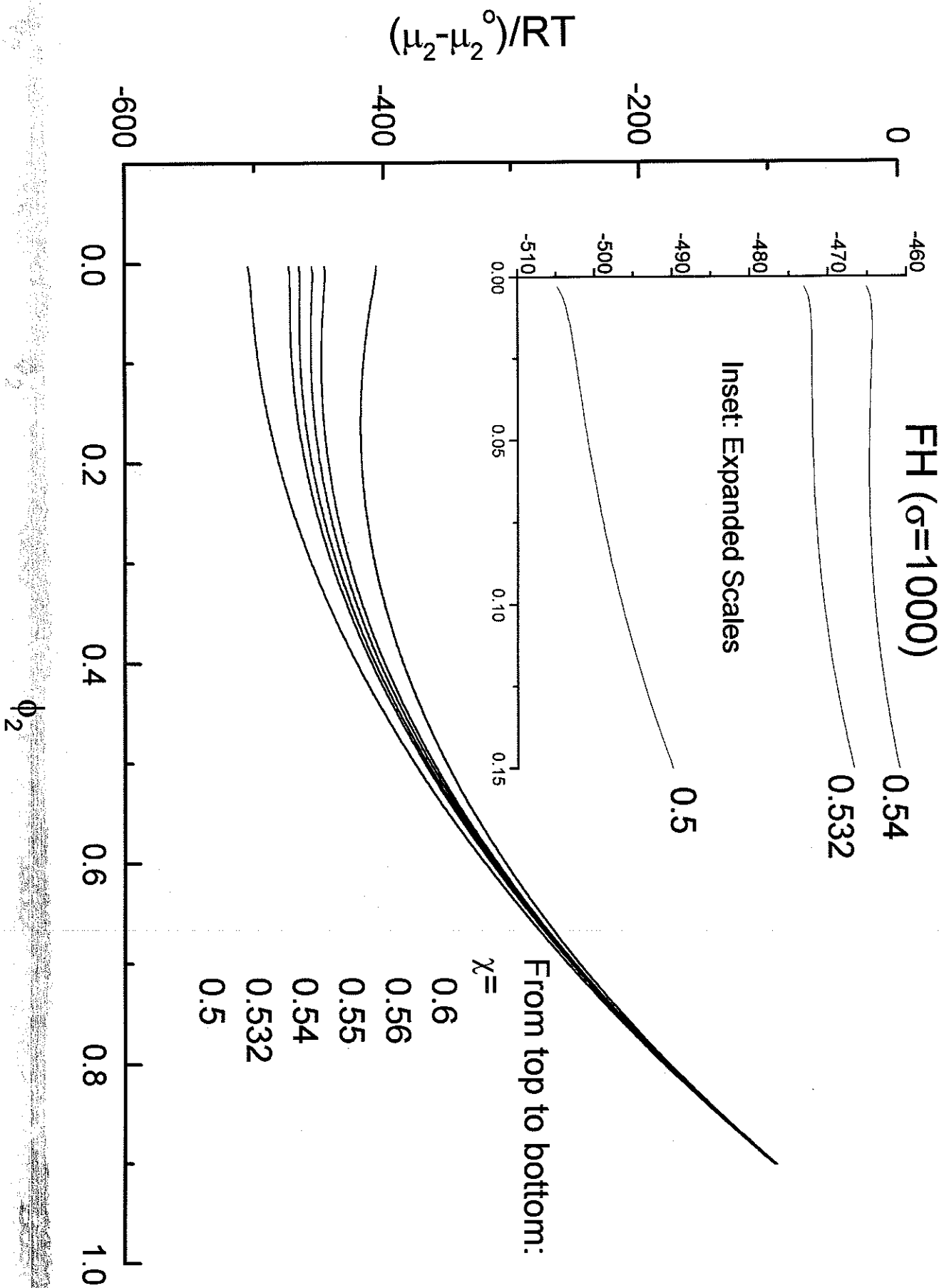
has the partner $\mu_2 - \mu_2^o = RT \left[\ln \phi_2 + (1 - \phi_2)(1 - \sigma) + X \sigma(1 - \phi_2)^2 \right]$

Phase Sep'n. occurs when 2 compositions exist at which μ_1 and μ_2 are the same in both phases.

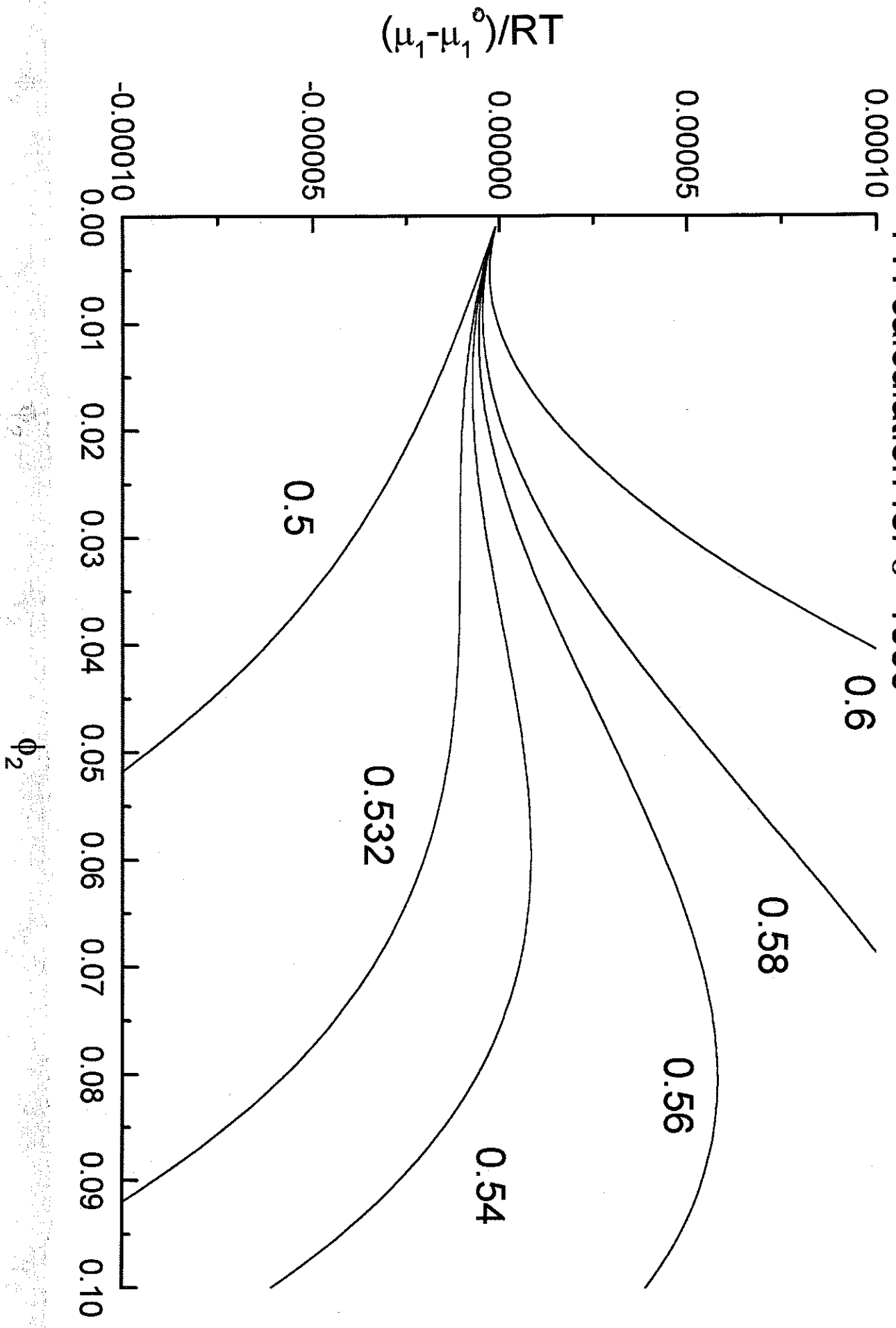
e.g. $\mu_1^a = \mu_1^b$ $\mu_2^a = \mu_2^b$

The following plots show that this happens only above a certain X_{crit} . Now we develop the theory.

★ why an exercise!

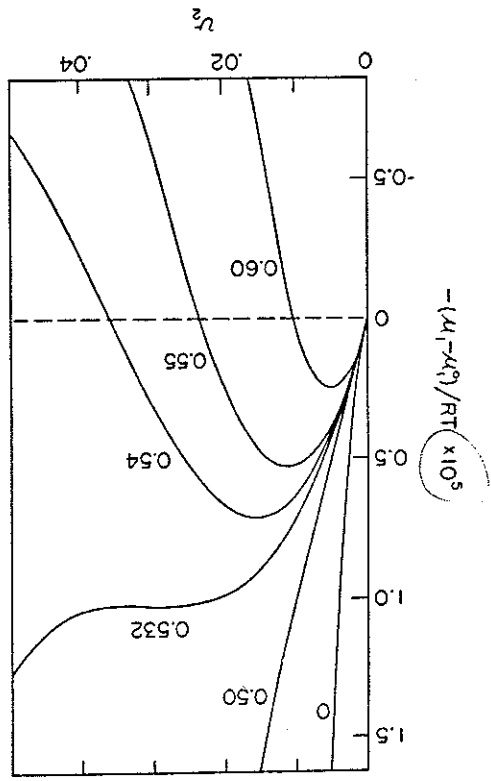


FH calculation for $\sigma=1000$



Del Ferry p. 574

Fig. 120.—The chemical potential of the solvent in a binary solution containing polymer at low concentrations (w_2). Curves have been calculated according to Eq. (XII-26) for $x = 1000$ and the values of x_1 indicated with each curve.



Can we find the critical condition?

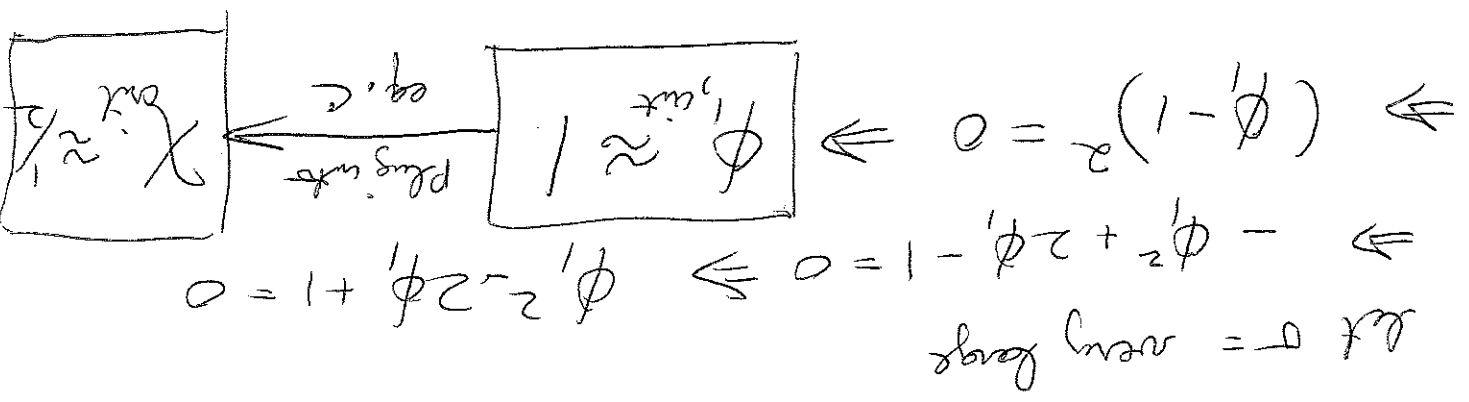
$$\mu_1 - \mu_1^0 = RT \ln \left[(1 - \phi_2) + \phi_2 \left(1 - \frac{1}{x} \right) + x \phi_2^2 \right]$$

Deriv of $\frac{\partial \mu_1}{\partial \phi_2} = 0 = \frac{\partial^2 \mu_1}{\partial \phi_2^2}$

(A) $\Rightarrow \frac{\partial \mu_1}{\partial \phi_2} = 0$
 $\frac{1}{1 - \phi_2} (-1) + \left(1 - \frac{1}{x} \right) + 2x \phi_2 = 0$

(B) $\frac{\partial^2 \mu_1}{\partial \phi_2^2} = 0$
 $\frac{-1}{(1 - \phi_2)^2} + 2x_{crit} = 0$

(C) $\Rightarrow x_{crit} = \frac{1}{2(1 - \phi_2)^2} = \frac{1}{2\phi_2^2}$



1ST GUESSES AT SOLUTION

MAYBE SKIP

$$\Rightarrow \phi_2 \left(\frac{\sigma}{1-\sigma} \right) + 2\phi_1 - 1 = 0$$

$$\boxed{\phi_2 = 1 - \phi_1}$$

$$\text{or } \dots \phi_2 \left(\frac{\sigma}{1-\sigma} \right) + \phi_1 - \phi_2 = 0$$

$$0 = \phi_1 - \phi_2 \left(\frac{\sigma}{1-\sigma} \right) - \phi_2 = 0$$

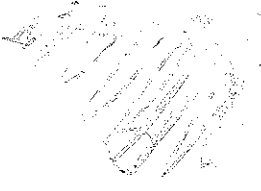
Modify eq 2

$$0 = \frac{\phi_1}{1} - \left(\frac{\sigma}{1-\sigma} \right) \phi_2 - \phi_2 = 0$$



$$0 = -\frac{1}{1} \phi_1 + \left(1 - \frac{\sigma}{1} \right) \phi_2 + \left(\frac{\sigma}{1} \right) \phi_2 = 0$$

Put @ into #



$$\boxed{\phi_{2, \text{crit}} = \frac{1-\sigma}{\sqrt{\sigma-1}}}$$

$$\phi_{2, \text{crit}} = 1 - \phi_{1, \text{crit}} = 1 - \frac{1-\sigma}{\sigma-\sqrt{\sigma}} = \frac{1-\sigma}{\sigma-1-\sigma+\sqrt{\sigma}}$$

$$\boxed{\phi_{1, \text{crit}} = \frac{\sigma-\sqrt{\sigma}}{\sigma-1}}$$

$\phi_{1, \text{crit}}$ cannot be > 1 , so choose \ominus term only

$$\phi_{1, \text{crit}} = \frac{1-\sigma}{\sigma \pm \sqrt{\sigma}} = \frac{1-\sigma}{\sigma-1}$$

$$= \frac{-2\sigma \pm 2\sigma\sqrt{\frac{\sigma}{1-\sigma}}}{2(1-\sigma)}$$

$$= \frac{-2 \pm 2\sqrt{1 + \frac{1-\sigma}{\sigma}}}{2(1-\sigma)}$$

$$\phi_{1, \text{crit}} = \frac{-2 \pm \sqrt{4 + 4(\frac{\sigma}{1-\sigma})}}{2(1-\sigma)}$$

$$\phi_2 \left(\frac{\sigma}{1-\sigma} \right) + 2\phi_1 - 1 = 0$$

In more exact form, we have in $(\frac{\sigma}{1-\sigma})$ and write:

* This equation doesn't exactly match Hiemenz because our eq. D is approximate.

$$\chi_{2,crit} = \frac{2\sigma}{(\sqrt{\sigma}+1)^2}$$

$$\chi_{crit} = \frac{2(1-\phi_{2,crit})^2}{1} = \frac{2(1-\frac{1}{\sqrt{\sigma+1}})^2}{1} = \frac{2\left(\frac{\sqrt{\sigma+1}}{\sqrt{\sigma+1}}\right)^2}{1}$$

a more accurate estimate:

if instead put $\phi_{2,crit} = \frac{1}{\sqrt{\sigma+1}}$ into (C) then you get

$$\chi_{crit} \approx \frac{2(1-\frac{1}{\sqrt{\sigma}})^2}{1} = \frac{2(\sqrt{\sigma}-1)^2}{\sigma} *$$

Put (D) into (C)

* easily remembered

$\phi_{2,crit} \approx \frac{1}{\sqrt{\sigma}}$

(D)

So...

$$\phi_{2,crit} = \frac{\sqrt{\sigma}-1}{(\sqrt{\sigma}-1)(\sqrt{\sigma}+1)} = \frac{1}{\sqrt{\sigma}+1}$$

But $\sigma-1 = (\sqrt{\sigma}-1)(\sqrt{\sigma}+1)$

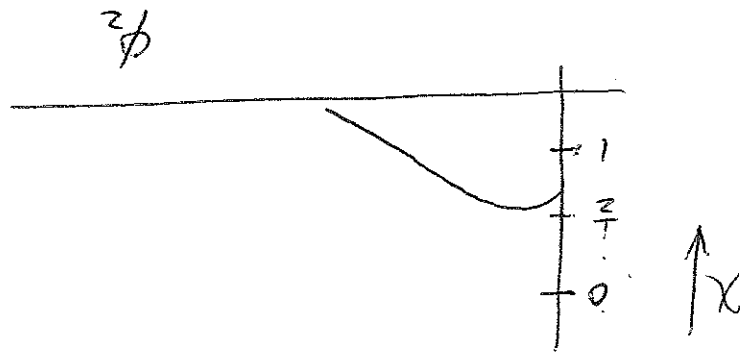
Now physically $\sigma \approx 100$ or more

So... $\phi_{2, \text{crit}} \approx 0$

$\chi_{\text{crit}} \approx \frac{1}{2}$

(Also! we predicted we would see $\text{No } \frac{1}{2}$ binomial!)

So... phase boundary is shifted to low ϕ_2



The great shift to low ϕ_2 matches experiment: a brilliant success.

Recall that $\chi \approx 0$ corresponds to good solvents

Polymers don't phase separate until χ becomes quite positive ($+\frac{1}{2}$). Reason is that it takes considerable "badness" to quench out all that configurational entropy. One could guess (correctly) that rigid polymers do not tolerate such high χ values - i.e. they phase separate at $\chi \approx 0$.

By inference, any real polymer (of real size & genuine stiffness) that has dissolved has done so in a good solvent.

Since χ_{crit} depends on T , you can fractionate polymers!

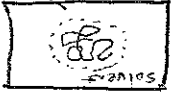
Recall we already did an α^2 calculation de Bruijn style. That calculation gave us no good solvent limit (i.e. $v = 3/5$) but made no connection to thermodynamics.

Experiment: any important problem can be solved several ways... one of which should be simple. No de Bruijn approach is simple. Now we do something a bit harder. We should again get $v = 3/5$ BUT also thermodynamics.

The following calculation is due to S. Flory. Some years after he did this (who notes, as always) in our polymer physics class, a paper appeared by another author: See E.A. Di Marzio, *Macromolecules* 1984, 17, 969. (as cited in the Flory & Leiberman text).

General idea: treat a single polymer coil as a small chunk of solution. Knowing the volume of that solution we connect $\langle r^2 \rangle$ to thermodynamics.

1) Imagine a polymer to be the very first one placed on a lattice.

2) After polymer is placed, let us put a confining balloon around it, or a loose "sac" 

3) Polymer must stay in the sac; solvent can permeate it.

On average, the dimensions of the polymer are those of the sac. But we chose the sac to have dimensions that are not too restrictive - i.e. about $\sqrt{\langle r^2 \rangle}$

Strategy

1) Calculate ΔG_{mix} for polymer constrained this way, as a function of sac

2) Minimize to find equilibrium

Compute # of configurations:

Consider Untrapped down

$$W \approx N_0^2 (z-1)^{z-2}$$

Assume $\mathcal{I}_I = 0$
 in Flory-Huggins calculation
 since this is first division
 lattice

Now, W counts ALL possible configurations, including some that won't fit in tray. We want a reduced

count, $W_r =$ # configurations subject to constraint that

This also ensures molecule is in an internally constrained
 length form that FH will work.
 printing configurations σ^2
 that are larger than σ^2

$$e^{-\frac{-3R^2}{2\sigma^2}} N_0^2 (z-1)^{z-2}$$

$$W_r = P(z) W$$

$$S_r = k_B \ln W_r = -\frac{3}{2} R^2 \frac{\sigma^2}{R^2} +$$

Usual Flory
 Huggins result
 which should be
 valid inside the domain
 as ϕ_2 , internal is large

(i.e., entropy is reduced because we constrain the molecule so it can't have all possible configurations but only those where arms and strand length = l_0)

$$\Delta G_{mix} / RT = \frac{3}{2} \ln \phi_2^2 + \left[N_1 \ln \phi_1 + N_2 \ln \phi_2 + X N_1 \phi_2^2 \right]$$

N_i are molecules not moles
 \Rightarrow This term is $\Delta G_{FLORY-HUGGINS} / RT$

More mixing of moles. $n = 2$

\Rightarrow FIRST TERM IS EASY

\Rightarrow 2ND TERM CAN BE DONE BY CHAIN RULE

$$\frac{\partial}{\partial N_1} \left(\frac{\partial N_1}{\partial r} \right) = \frac{\partial}{\partial N_1}$$

Taking $\frac{\partial}{\partial N_1} \left(\frac{\Delta G}{RT} \right)$ will give same

Answer as $\frac{\partial}{\partial m_1} \left(\frac{\Delta G_{FH}}{RT} \right)$ which we obtained

already: $\ln \phi_1 = \ln \left(1 - \phi_2 \right) + \phi_2 \left(1 - \frac{1}{r} \right) + X \phi_2^2$

$$\Rightarrow 0 = \frac{3}{2} \ln \phi_2^2 + \left(\frac{\partial N_1}{\partial r} \right) \left(\ln \left(1 - \phi_2 \right) + \phi_2 \left(1 - \frac{1}{r} \right) + X \phi_2^2 \right)$$

$$V_{sac} \approx \frac{4}{3} \pi R^3$$

$$V_{of\ solvent} = (1 - \phi_2) V_{sac}$$

$$\Rightarrow N_1 = \frac{(1 - \phi_2)^{4/3} \pi R^3}{(V_1^0 / V)}$$

$\frac{Volume\ of\ solvent\ in\ sac}{Volume\ per\ molecule\ of\ solvent}$

$$0 \approx \frac{3R}{2} + 9\sigma \left(\frac{V_0/N}{4\pi R^3} \right) \left(x - \frac{1}{2} \right) - \frac{2}{R}$$

Now use $\phi_2 = \frac{\sigma(V_0/N)}{4\pi R^3}$ again & expand

$$0 \approx \frac{3R}{2} + \frac{3\sigma}{R} \left[\phi_2 \left(x - \frac{1}{2} \right) - \frac{\sigma}{R} \right]$$

$$\therefore 0 \approx \frac{3R}{2} + \frac{3\sigma}{R} \left[\phi_2 \left\{ \phi_2 \left(x - \frac{1}{2} \right) - \frac{\sigma}{R} \right\} \right]$$

Therefore: $\frac{4\pi R^2}{3\sigma} = \frac{(V_0/N)}{\phi_2}$

Now $\phi_2 = \frac{\frac{4}{3}\pi R^3}{\text{volume of } \sigma \text{ polymer segments}} = \frac{\frac{4}{3}\pi R^3}{\text{volume of } \sigma}$

$$0 \approx \frac{3R}{2} + \frac{\sigma(V_0/N)}{4\pi R^2} \left[-\cancel{\phi_2} - \frac{\phi_2}{2} + \cancel{\phi_2} - \frac{\sigma}{R} + x\phi_2 \right]$$

$\underbrace{\phi_2 \left\{ \phi_2 \left(x - \frac{1}{2} \right) - \frac{\sigma}{R} \right\}}_{\text{from above}}$

$$\therefore 0 = \frac{3R}{2} + \frac{\sigma(V_0/N)}{4\pi R^2} \left[\ln(1-\phi_2) + \phi_2 \left(1 - \frac{1}{2} \right) + x\phi_2 \right]$$

Thus $\frac{\partial N}{\partial R} = \frac{(1-\phi_2) 4\pi R^2}{(V_0/N)} \approx \frac{V_0/N}{4\pi R^2}$

$$\begin{aligned} \alpha &= 2 \\ \frac{r_3}{r_5} &= \frac{32}{8} = 4 \\ \frac{r_3}{r_5} &= 3 \\ \frac{r_3}{r_5} &= \frac{243}{27} = 9 \end{aligned}$$

$$\alpha \sim \sigma^{1/10} = \sigma^{0.1}$$

consistent with de Vries' "Spring" approach

OBSERVATION #1
At large σ , $\alpha^5 \gg \alpha^3$, so...

about "3" $\Rightarrow K' \sigma^{1/2} (\frac{1}{2} - \alpha)$ and α^3

$$\alpha^5 - \alpha^3 = -K \sigma^2 (\alpha - \frac{1}{2}) \Rightarrow \text{by both sides by } 3(\sigma \alpha^2)^{3/2}$$

$$\Rightarrow 3 \alpha^5 (\sigma \alpha^2)^{3/2} + K \sigma^2 (\alpha - \frac{1}{2}) - 3 \alpha (\sigma \alpha^2)^{3/2} = 0$$

Now $r_2 = \alpha^2 \sigma \alpha^2$
 $r_5 = \alpha^5 (\sigma \alpha^2)^{5/2}$
 $r_3 = \alpha^3 (\sigma \alpha^2)^{3/2}$

$$\Rightarrow 0 = \frac{3r_5}{\sigma \alpha^2} + K \sigma^2 (\alpha - \frac{1}{2}) - 3r_3$$

$$\frac{9}{4\pi N}$$

Remove all r 's from denominator (mult. by r^4)

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- The first big failure of FH

Turn out that experiments show $A_2 \sim M^{-0.2}$

Note: this predicts A_2 independent of M .
 Recall that this is the same as rods, while spheres gave $A_2 = \frac{4\sqrt{2}}{3} M^{-1}$

$$\frac{\pi}{c} = RT \left[\frac{1}{M} + \left(\frac{z}{2} - \chi\right) \frac{V_0}{M^2} c + \dots \right]$$

As usual, use $\pi = -(p - p_0)$

Note that $\chi = 0$, A_2 is positive: repulsion
 $\chi \ll 0$, A_2 gets large: ionic interaction
 in polyelectrolytes
 might cause
 them

So $A_2 = \left(\frac{V_0}{M^2}\right)^2 \left(\frac{z}{2} - \chi\right) = 0$ @ $\chi = \frac{z}{2}$

and $\frac{V_0}{M^2} \approx \frac{M_2}{M^2}$

Now $\sigma V_0 = V_2$

Thus, $A_2 = \sigma^2 \frac{M_2^2}{M^2} \left(\frac{z}{2} - \chi\right)$